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# Cytotoxic, Antifouling Bromotyramines: A Synthetic Study on Simple Marine Natural Products and Their Analogues

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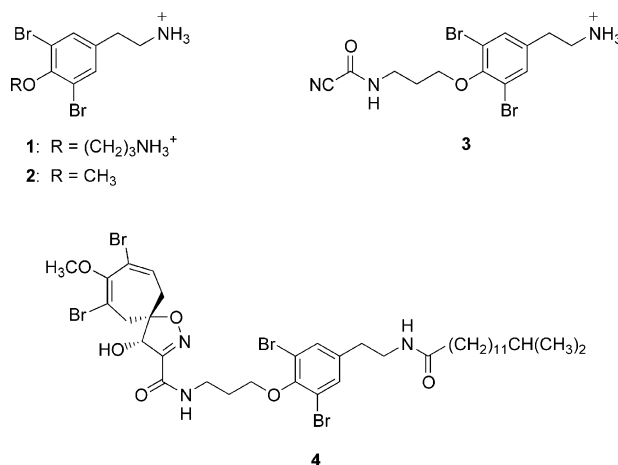
**Abstract**—Synthesis and biological evaluation of two naturally-occurring bromotyramines, moloka'iamine **1** and 3,5-dibromo-4-methoxy- $\beta$ -phenethylamine **2**, together with several analogues, have been completed. Bromotyramine **2** is cytotoxic, and was found to be a potent antifoulant. Analogues **15** and **16** also displayed significant cytotoxic and antifouling activities. © 2002 Elsevier Science Ltd. All rights reserved.

Marine organisms are a rich source of chemically diverse bioactive substances. Over the past three decades, bromotyrosine-derived natural products have frequently been isolated from sponges of the order Verongida,<sup>1</sup> as well as from a few other marine sources such as ascidians.<sup>2</sup> Those compounds exhibit antifouling, antiviral, cytotoxicity, antifungal, and antibacterial activities.<sup>3</sup> Moloka'iamine (**1**), a compound first isolated in 1993 from a Verongid sponge of unidentified species,<sup>4</sup> is based on an *O*-alkylated dibromotyramine core, and displays cytotoxic and antifouling activity.<sup>5</sup> The *O*-alkylated dibromotyramine motif can be found in nature in both large, structurally complex natural products such as ceratinamide **B** (**4**),<sup>6</sup> as well as in its simplest form, that is 3,5-dibromo-4-methoxy- $\beta$ -phenethylamine (**2**) (Fig. 1)<sup>7</sup>

The antifouling activity of **1** against barnacle cyprids of *Balanus amphitrite* was particularly noteworthy.<sup>5</sup> Natural product antifoulants have been extensively investigated recently in search of environmentally benign replacements for tributyltin in maritime coatings applications.<sup>8</sup> We felt that the relatively simple structure of **1** invited the synthesis of analogues in an effort to probe antifouling structure–activity relationships. Since **1** was also reported to be active against P388 murine leukemia

cells ( $IC_{50}$  = 2.1  $\mu$ g/mL),<sup>5</sup> further antitumor and cytotoxic screening also seemed warranted.

We previously reported the synthesis of **1** in conjunction with the total synthesis of ceratinamine **3**,<sup>9</sup> a functionalized bromotyramine exhibiting potent antifouling and cytotoxic activities.<sup>5</sup> For the present study, a new synthetic plan was devised that would allow for the convenient preparation of **1** and its analogues on a suitably

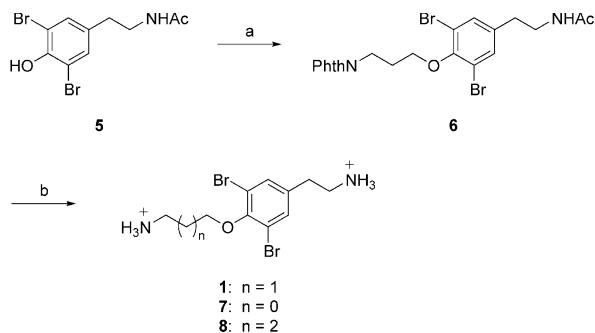


**Figure 1.** Bromotyramine marine natural products moloka'iamine (**1**), 3,5-dibromo-4-methoxy- $\beta$ -phenethylamine (**2**), ceratinamine (**3**), and ceratinamide B (**4**).

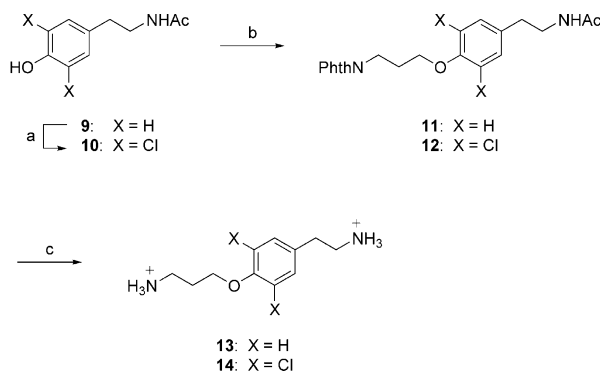
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large scale (Scheme 1). Reaction of *N*-acetyl-3,5-dibromo-4-hydroxy- $\beta$ -phenethylamine **5**<sup>10</sup> with 3-bromopropylphthalimide, potassium carbonate, and catalytic potassium iodide in acetonitrile at reflux smoothly afforded protected diamine **6** in 98% yield. Exhaustive hydrolysis with concentrated HCl at reflux furnished moloka'iamine **1** as its dihydrochloride salt. The ethyl and *n*-butyl analogues **7** and **8** were prepared in similar fashion.

*N*-Acetyltyramine **9**<sup>11</sup> was used as starting material for the synthesis of additional analogues (Scheme 2). Chlorination of **9** with sulfuryl chloride in ether afforded **10** in modest yield. Phenols **9** and **10** were then



**Scheme 1.** Reagents and conditions: (a) PhthN(CH<sub>2</sub>)<sub>3</sub>Br, K<sub>2</sub>CO<sub>3</sub>, KI, acetonitrile, reflux, 98%; (b) concd HCl, reflux, 84%.



**Scheme 2.** Reagents and conditions: (a) SO<sub>2</sub>Cl<sub>2</sub>, diethyl ether, 38%; (b) PhthN(CH<sub>2</sub>)<sub>3</sub>Br, K<sub>2</sub>CO<sub>3</sub>, KI, acetonitrile, reflux, 87–98%; (c) concd HCl, reflux, 21–86%.

**Table 1.** Antifouling activity of bromotyramine natural products (**1** and **2**) and their analogues (**7–8** and **13–16**) against barnacle *Balanus amphitrite*

Compd	Settlement inhibition EC <sub>50</sub> (μg/mL)	Lethality LD <sub>50</sub> (μg/mL) <sup>a</sup>
<b>1</b>	5.0	nt
<b>2</b>	0.07	0.2
<b>7</b>	0.8	nt
<b>8</b>	6.0	nt
<b>13</b>	> 50	nt
<b>14</b>	33	nt
<b>15</b>	0.2	1.0
<b>16</b>	0.008	0.03

nt, not tested.

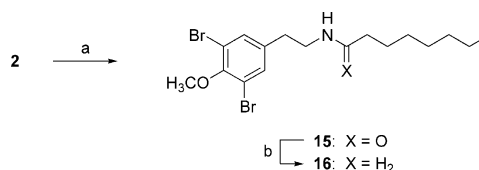
<sup>a</sup>Concentrations at which the compounds tested were lethal to 50% of the barnacle cyprids.

aminopropylated and deprotected as described above in the synthesis of **1** to give the unhalogenated and chlorinated analogues **13** and **14**, respectively.

Initially, **1** and its analogues **7**, **8**, **13**, and **14** were assayed against *B. amphitrite* using a previously described cyprid settlement assay (Table 1).<sup>12</sup> The weak activity observed with **13** and **14** indicated that bromines were strictly required for antifouling performance. The most active compound tested was the ethyl analogue **7**.

Noting the increased activity associated with a shorter alkyl chain, we decided to investigate the naturally-occurring bromotyramine **2**, in which a methyl group replaces the aminopropyl chain found in **1**. Compound **2**, first synthesized in 1958 as part of a veterinary clinical study<sup>10</sup> and more recently isolated from an *Eudistoma* sp. Ascidian,<sup>7</sup> was found to be the most potent antifouling bromotyramine reported to date, a full two orders of magnitude more active than **1**.

Bromotyramines **15** and **16** were designed to hybridize the aromatic portion of **2** with the aliphatic portion of an antifouling natural product such as ceratinamide B (**4**, IC<sub>50</sub> = 2.4 μg/mL vs *B. amphitrite*),<sup>6</sup> having a long-chain alkyl group appended to the phenethylamine nitrogen. We reasoned that a more lipophilic version of **2** would be more soluble in a coating, and therefore more practical as an antifouling paint additive. Acetylation of **2** with octanoyl chloride afforded amide **15** in near-quantitative yield (Scheme 3). Reduction of **15** with borane-THF, followed by an acidic workup, furnished octylamine **16** as its HCl salt. Compounds **15** and **16** strongly inhibited the settlement of barnacle cyprids, with IC<sub>50</sub> values of 0.2 and 0.008 μg/mL, respectively. It



**Scheme 3.** Reagents and conditions: (a) C<sub>7</sub>H<sub>15</sub>COCl, triethylamine, methylene chloride, 99%; (b) BH<sub>3</sub>-THF, 98%.

**Table 2.** Growth inhibition of human cancer cell lines by selected bromotyramine compounds (**1**, **2**, **3**, **5**, **7**, **8**, **13**, **14**, **15**, and **16**) at 100 μM

Compd	% Growth		
	NCI-H460 (lung)	MCF-7 (breast)	SF-268 (CNS)
<b>1</b>	68	38	62
<b>2</b>	53	8	16
<b>3<sup>a</sup></b>	82	7	39
<b>5</b>	105	88	84
<b>7</b>	70	39	70
<b>8</b>	25	19	50
<b>13</b>	120	104	120
<b>14</b>	93	–53	85
<b>15</b>	1	–74	–68
<b>16</b>	–83	–82	–84

<sup>a</sup>Compound **3** (ceratinamine) was synthesized as described previously.<sup>9</sup>

should be noted that bromotyramines **2**, **15**, and **16** exhibited varying levels of cyprid toxicity (Table 1), to which the observed antisettlement activity may well be ascribed. Investigation into the use of **16** as an anti-fouling paint additive is being pursued.

Some of the compounds described herein were submitted to the US National Cancer Institute for screening against several human tumor cell lines.<sup>13</sup> Prescreening results (Table 2) indicated that compounds **2**, **15**, and **16** demonstrated sufficient activity to pass to the 60-cell line screen. All three compounds displayed cytotoxicity at a mean panel GI<sub>50</sub> concentration of 10 μM. Bromotyramine **2** was especially active in the CCRF-CEM leukemia, NCI-H226 lung cancer, and SW-620 colon cancer cell lines (GI<sub>50</sub> <0.01 μM for each), as well as in the Hs578T breast cancer cell line (GI<sub>50</sub>=0.2 μM). Compound **15** was most active in the CCRF-CEM leukemia cell line (GI<sub>50</sub>=0.4 μM), while **16** displayed the greatest activity in the NCI-H460 lung cancer cell line (GI<sub>50</sub>=0.05 μM).

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8018643) and the National Institutes of Health (RR02002) is gratefully acknowledged.

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